STRUCTURE-ACTIVITY RELATIONSHIPS OF THE PYRIDAZINONE SERIES OF 5-LIPOXYGENASE INHIBITORS

Dee W. Brooks*, Anwer Basha, Francis A. J. Kerdesky, James H. Holms, James D. Ratajcyk,
Pramila Bhatia, Jimmie L. Moore, Jonathan G. Martin, Steven P. Schmidt,
Daniel H. Albert, Richard D. Dyer, Patrick Young, George W. Carter
Immunosciences Research Area, Department 47K, Abbott Laboratories, Abbott Park, Illinois 60064
(Received 30 June 1992)

Abstract. Structure activity analysis of the pyridazinone series as represented by the initial lead compound A-53612 revealed that the 1-phenyl-2H-tetrahydropyridazin-3-one structure was necessary for inhibitory activity as several modifications diverging from this structure led to a dramatic loss of inhibitory activity. Substituents on the phenyl ring had a marked effect on inhibitory activity and methemoglobinemia toxicity.

Inhibition of leukotriene biosynthesis by inhibition of the enzyme 5-lipoxygenase (5-LO) represents a promising new therapeutic intervention.¹ The discovery of A-53612, 1-phenyl-[2H]-tetrahydropyridazin-3-one as a selective orally active 5-LO inhibitor provided a lead compound for further optimization.² This report describes the structure-activity relationships of this series of 5-LO inhibitors.

Chemistry. The syntheses of the novel compounds evaluated in this investigation are described as follows.³ The phenyl substituted pyridazinones, 1-21 were prepared by either Method A: condensation of the appropriate substituted acetylphenylhydrazide I with ethyl 4-bromobutyrate in the presence of ethyldiisopropylamine followed by treatment with sodium ethoxide in ethanol to provide the pyridazinone II or Method B: heating the corresponding substituted aniline III with methyl 4-bromobutyrate in the presence of diisopropylethylamine followed by nitrosation and reduction to the intermediate hydrazine III which was cyclized by treatment with sodium ethoxide in ethanol to yield the desired pyridazinone II. The carboxyl analog 22 was prepared by saponification of the ester 20. The 1-phenylpyrazolidinones 23-29 were prepared by known methods⁹ involving condensation of the requisite phenylhydrazine with acrylonitrile followed by acid hydrolysis.

Compounds 30-32 were prepared by treatment of 1 with methyliodide, propionylchloride, and benzoylchloride, respectively. The cyclohexyl analog 34 was prepared by catalytic hydrogenation of 1 using 5% rhodium on alumina as catalyst. Compounds 35-39 were prepared by reaction of the hydrazide 52 with pentylbromide, benzoylchloride, octylbromide, and dodecylbromide, respectively.

The novel seven- and eight-membered hydrazides 41 and 43 were prepared by intramolecular cyclization of the corresponding hydrazine esters 40 and 42 by treatment with trimethylaluminum. Hydrazines 40 and 42 were prepared from aniline and the appropriate *omega*-bromo ester by Method B.

Identification of the Pharmacophore for 5-Lipoxygenase Inhibitory Activity. A survey of the structural requirements for inhibitory activity was conducted using the 5-lipoxygenase activity in the 20,000xg supernatant from sonicated RBL-1 cells.⁴ Several structural modifications described as follows supported the premise that 1-phenyl-2H-tetrahydropyridazin-3-one (1) was the pharmacophore for 5-LO inhibitory activity. The following structural modifications resulted in inactive compounds: 1) removal of the carbonyl function as in 46^5 and 47^6 , 2) replacement of the nitrogen atom at position 1 with carbon as in 48^7 and 49^8 , and 3) at position 2 as in 50^9 . A carbonyl at position 6 as in 51^{10} provided weak activity (24% at 30 μ M).

The phenyl group at N1 proved to be essential for 5-LO inhibitory activity as the following analogs were inactive: 1) removal of the phenyl group as in 2H-tetrahydropyridazinone (52)¹¹, 2) repositioning the phenyl group at carbon 6 as in 53^{12} and 54, and 3) placing the phenyl group at N2 as in 55^{13} and 56^{14} .

Substitution on Nitrogen. N2-alkyl analogs such as N-methyl 30, N-phenyl 33¹⁵ and N2-acyl analogs 31 and 32 were inactive. Thus, an unsubstituted 2-NH group was required for 5-LO inhibitory activity.

Replacement of the Phenyl Group. Replacement of the phenyl group at N1 was investigated as follows. The 1-cyclohexyl congener 34 and the 1-n-pentyl analog 35 were both inactive. The benzyl analog 36 and the benzoyl analog 37 were inactive. Appending a lipophilic hydrocarbon chain at N1, as in the octyl 38 and dodecyl analogs 39, provided modest inhibitors, $IC50'_{S} = 23$ and 6 μ M, respectively.

Effect of Ring Size and Conformation. The effect of the ring size of the cyclic hydrazide was examined. The five-membered analog, phenidone had similar *in vitro* activity as the six-membered analog, A-53612². Surprisingly, the seven-membered analog 41 was devoid of 5-lipoxygenase inhibitory activity at concentrations less than 32 μ M and the eight-membered analog 43 was marginally active, 8% at 32 μ M. This dramatic difference in inhibitory activity indicated that the conformational configuration of the cyclic N-phenylhydrazide unit was important for effective inhibition of 5-LO. The conformation requirements of the hydrazide unit were further explored by examining simple acyclic phenylhydrazides such as 44 (IC₅₀ = 65 μ M) and 45 (IC₅₀ = 52 μ M) which showed more than a ten-fold reduction in inhibitory activity compared to A-53612 (1). On the basis of this survey of structural modifications we concluded that A-53612 was a minimal structural unit for satisfactory 5-lipoxygenase inhibitory activity.

Substitution on the Phenyl Group. The structure-activity relationships of the pyridazinone series with a variety of substituents on the phenyl ring were evaluated. A very interesting pattern of inhibitory activity was observed and the results are summarized in Table 1. Substitution in the *ortho* or *para* positions resulted in a significant loss of inhibitory activity as illustrated for methyl 2, 4; ethyl 5, 7; chloro 8, 10 and methoxy 11, 13 derivatives. Similar substitution with these same groups at the *meta* position 3, 6, 9, 12 only slightly modulated the inhibitory activity compared to the parent system 1. This phenomenon is in contrast to that observed for the substituted five-membered phenidone system, shown in Table 2. As demonstrated for the methyl 24-26 and chloro 27-29 substituted phenidone analogs, the position of the substituent did not change inhibitory activity as dramatically as in the pyridazinone system. The pyridazinone series was much more sensitive to substituent effects which suggests a more specific binding and alignment with the enzyme.

Further analysis of various substituents at the *meta* position in the pyridazinone series was conducted in a effort to maximize inhibitory activity (Table 1). Only small improvements in inhibitory activity were found with the methyl 3, ethyl 6, chloro 9, bromo 14 derivatives compared to the parent 1. The more polar substituents resulted in pronounced loss of inhibitory activity.

Structure-Activity Relationships With Respect to Methemoglobinemia. The removal of the toxicity associated with the methemoglobinemia expressed by A-53612 in vivo 2 was an important objective in order to identify a suitable therapeutic agent. Several analogs were evaluated by daily oral doses of 400 mg/kg in rats and measuring the proportion of red blood cells containing Heinz bodies. Of the active pyridazinone inhibitors, the nitrile analog 18 (A-63640) was unique as it exhibited no Heinz bodies after 11 days at 400 mg/kg per day. The plasma level of 18 achieved in the rat after oral dosing was similar to 1 (A-53612). However this inhibitor did not have adequate *in vivo* potency in the rat anaphylaxis assay 16 (ED₅₀ = 23 mg/kg) to merit further development.

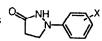
Summary. A-53612, 1-phenyl-2H-tetrahydropyridazin-3-one (1) offered a lead for the development of selective, orally active 5-LO inhibitors. The position of substituents on the phenyl ring had a marked effect on inhibitory activity indicating a specific binding interaction with the enzyme. Substituents in the *ortho* or *para* positions resulted in a loss of inhibitory activity whereas non-polar substituents in the *meta* position had minor influence on inhibitory activity. The observation that A-63640 (18) did not cause methemoglobinemia after eleven daily oral doses provided support that this toxic effect could be overcome in this series. The SAR elucidated in this study indicated a narrow range of productive modifications for further optimization of potency.

Table 1. Substituted 1-Phenyl-[2H]-tetrahydropyridazin-3-ones

Compound	X	IC ₅₀ ,a μΜ	Compound	i x ^	IC_{50} ,a	μM
1	Н	3.8 (3.1-4.6)	12	3 - OCH ₃	6.7	(6.1-7.4)
2	2 - CH ₃	57 (56-58)	13	4 - OCH ₃	156	(140-173)
3	3 - CH ₃	3.4 (2.8-4.0)	14	3 - Br	1.8	(1.6-2.1)
4	4 - CH ₃	277 (261-294)	15	3 - CH ₂ OH	53	(45-60)
5	2 - CH ₂ CH ₃	39 (33-47)	16	3 - COCH ₃	19	(16-21)
6	3 - CH ₂ CH ₃	3.0 (2.5-3.8)	17	3 - COtC4H9	15	(13-17)
7	4 - CH ₂ CH ₃	125 (116-136)	18	3 - CN	9.3	(8.1-11)
8	2 - Cl	99 (90-108)	19	3 - CF ₃	5.2	(4.6-5.7)
9	3 - Cl	2.5 (2.2-3.0)	20	3 - COOCH2CH3	5.6	(4.9-6.6)
10	4 - Cl	79 (60-99)	21	3 - F	4.5	(4.2-4.8)
11	2 - OCH ₃	204 (176-237)	22	3 - COOH	NA b up t	ο 300 μΜ

a. IC50 with 95% confidence limits indicated in parentheses. b. NA indicates no significant inhibitory activity.

Table 2. Substituted 1-Phenylpyrazolidin-3-ones



Compound	X	IC ₅₀ a, μΜ	Compound	X	IC ₅₀ a, μΜ
23	Н	1.9 (1.5-2.8)			
24	2 - CH3	8.2 (7.2-9.4)	27	2 - Ci	3.7 (3.4-4.1)
25	3 - CH ₃	1.3 (1.0-1.7)	28	3 - Cl	0.73 (0.62-0.87)
26	4 - CH ₃	3.5 (3.1-3.9)	29	4 - Cl	1.1 (0.86-1.4)

a. IC₅₀ with 95% confidence limits indicated in parentheses.

References

- 1. Lewis, R.A.; Austen, F.; Soberman, R.J. New England J. of Medicine 1990, 323, 645.
- 2. Brooks, D.W.; Albert, D.H.; Dyer, R.D.; Bouska, J.B.; Young, P; Carter, G.W. preceeding article.
- 3. Detailed procedures for the preparation of the active inhibitors are reported in: Brooks, D.W.; Carter, G.W.; Basha, A.; Gunn, B.P.; Dyer R.D. U.S. Patent 5,086,052.
- The term inactive is defined as no significant inhibition up to 30 μM in the 5-LO inhibition assay reported by Dyer, R. D.; Bornemeier, D. A.; Haviv, F.; Carter, G. W. Fed. Proc., Fed. Am. Soc. Exp. Biol. 1985, 44, 904.
- 5. Eberle, M.K.; Kahle, G. G.; Talati, S. M. Tetrahedron 1973, 29, 4045.
- 6. Aubaynac, J. L.; Elguero, J.; Jacquier, R.; Robert, R. C. R. Acad. Sci., Ser. C, 1970, 270, 1829.
- 7. DeGraaff, B. R.; Melger, W. C.; Van de Kolk, G. Rec. Trav. Chim. 1962, 81, 786.
- 8. Diller, D.; Bergmann, F. J. Org. Chem. 1972, 37, 2147.
- 9. Taylor, E. C., Skotnicki, J. S. Synthesis 1981, 606.
- 10. Pirkle, W. H., Gravel, P.L. J. Org. Chem. 1977, 42, 1367.
- 11. Gut, J., Novacek, A. Fiedler, P. Collect. Czech. Chem. Commun. 1968, 33, 2087.
- 12. Delaby, R. Chabrier, P. Danton S. Compt. Rend. 1953, 66, 237.
- 13. Lui, H.; Warkentin, J. Can J. Chem. 1972, 59, 1767.
- Ege, S.; Carter, M.L.C.; Ortwine, D.F.; Chow, S.; Shing, P.; Richman, J.F. J. Chem. Soc. Perkin Trans. I, 1977, 11, 1252.
- 15. Degrand, C; Jacquin, D. Tetrahedron Letters 1978, 4955.
- The rat anaphylaxis model is described in detail in: Young, P.R.; Bell, R.L.; Lanni, C.; Summers, J.B.; Brooks, D.W.; Carter, G.W. Eur. J. Pharm. 1991, 205, 259.